TOTAL MATERIAL REQUIRMENT

and its Revaluation as as a Resource Frootprint

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ΓMR

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TOTOE NOIVERE ROOODD



| chemicals | TCFP | TMR | chemicals | TCFP | TMR | chemicals | TCFP | TMR | chemicals | TCFP | TMR | chemicals | TCFP | TMR |
|---------------------------------|----------|----------|-----------------------------------|----------|-----------|-------------------------------|----------|----------|---------------------------------|----------|----------|----------------------------------|----------|------------|
| l,4-Butanediol | 7.87E+01 | 8.93E+02 | chilesin | 1.19E+01 | 1.54E+02 | hexane | 4.75E+00 | 5.05E+01 | nitrogen (N) | 5.71E-01 | 7.95E+00 | sodium chloride | 1.04E+0 | 1 9.20E+01 |
| 2-Ethylene hexanol | 8.71E+03 | 1.28E+05 | chlorine (Cl) | 2.08E+01 | 1.76E+02 | High Purity Quartz | 1.20E+02 | 1.35E+03 | o Tetrahydrofuran | 4.97E+01 | 5.53E+02 | sodium cyanide | 7.25E+00 | 0 7.77E+01 |
| acetone | 2.38E+01 | 3.29E+02 | chromium (Cr) | 1.47E+02 | 2.15E+03 | High-purity silicon | 4.09E+01 | 4.05E+02 | ocean water | 0.00E+00 | 1.00E+00 | sodium fluoride (NaF) | 5.62E+0 | 2 6.21E+03 |
| acetylene | 6.22E+01 | 7.03E+02 | chromium concentrates | 1.91E+01 | 4.31E+02 | holmium oxide (HoO) | 2.19E+06 | 2.89E+07 | ore dressing | 1.57E+01 | 3.47E+03 | sodium hydroxide | 1.89E+0 | 1 1.59E+02 |
| activated charcoal | 5.62E+00 | 4.10E+01 | chromium nitrate | 8.43E+01 | 1.69E+03 | hydrochloric acid | 1.01E+01 | 9.01E+01 | Ore Mining | 2.71E+00 | 3.99E+01 | sodium hypochlorite | 2.46E+0 | 1 2.05E+02 |
| agrochemical | 1.75E+01 | 5.18E+02 | Cinnabar concentrates | 2.55E+01 | 3.64E+02 | hydrogen | 6.00E+01 | 7.05E+02 | Organic liquid waste treatment | 1.21E+00 | 1.05E+01 | sodium lauryl sulfate (sodium la | 4.00E+0 | 1 5.39E+03 |
| air | 0.00E+00 | 1.00E+00 | coagulant | 1.31E+01 | 1.11E+02 | hydrogen fluoride | 6.13E+02 | 6.79E+03 | osmium (Os) | 1.88E+07 | 2.77E+08 | sorbitan | 4.62E+0 | 1 2.19E+03 |
| alcohol | 4.22E+00 | 1.72E+03 | coal | 3.72E-01 | 9.59E+00 | hydrogen peroxide | 2.56E+00 | 2.25E+01 | oxidized lantern | 7.86E+03 | 1.04E+05 | sorbitol (artificial sweetener) | 3.45E+0 | 1 1.90E+03 |
| allylbenzene | 5.17E+00 | 5.60E+01 | cobalt (Co) | 6.29E+02 | 8.52E+03 | ilmenite concentrate | 1.93E+01 | 2.55E+02 | oxygen | 6.38E-01 | 9.99E+00 | stearic acid | 2.44E+0 | 1 2.75E+02 |
| alumina | 5.14E+00 | 1.64E+02 | cobalt concentrate | 4.52E+01 | 6.38E+02 | indium (In) | 5.21E+04 | 8.37E+05 | palladium (Pd) | 1.03E+05 | 1.44E+06 | strontium carbonate (SrCO3) | 5.79E+0 | 1 7.29E+02 |
| aluminium fluoride | 1.09E+03 | 1.21E+04 | Coconut Fruit | 2.58E+00 | 9.19E+02 | Industrial Carbon | 2.74E+01 | 3.52E+02 | petroleum coke | 7.32E+00 | 9.91E+01 | sulfur (S) | 6.06E+0/ | 0 6.86E+01 |
| aluminum (Al) | 1.76E+02 | 2.14E+03 | coconut oil | 1.57E+01 | 4.52E+03 | Industrial silicon | 7.25E+01 | 8.12E+02 | PGM concentrate | 3.84E+04 | 5.65E+05 | sulfuric acid | 2.57E+0 | 0 2.51E+01 |
| aluminum chloride | 2.35E+01 | 2.54E+02 | coke (carbon fuel) | 2.51E+00 | 2.46E+01 | iodine (I) | 7.76E+03 | 3.50E+05 | phosphate ore | 1.23E+01 | 1.77E+02 | Sulfurous acid gas treatment | 8.84E+0 | 0 9.84E+01 |
| ammonia | 1.16E+01 | 1.41E+02 | copper | 3.44E+02 | 4.89E+03 | iridium (Ir) | 2.03E+07 | 2.99E+08 | phosphoric acid (H3PO4) | 7.30E+00 | 7.54E+01 | surfactant | 1.22E+0 | 1 1.23E+02 |
| ammonium oxide | 1.22E+01 | 1.36E+02 | Copper anode mud | 2.66E+03 | 3.82E+04 | iron | 1.06E+01 | 6.55E+01 | phosphorus | 1.06E+02 | 1.44E+03 | tantalum (Ta) | 6.86E+0 | 2 8.73E+03 |
| ammonium sulfate | 8.63E+00 | 8.51E+01 | Copper catalyst | 3.98E+02 | 5.72E+03 | iron ore | 1.01E+01 | 1.60E+02 | PHS | 9.27E+00 | 1.32E+02 | Tantalum concentrate | 4.82E+0 | 1 6.66E+02 |
| antimony (Sb) | 7.42E+01 | 9.20E+02 | copper concentrate | 9.68E+01 | 1.39E+03 | iron powder | 2.66E+00 | 1.06E+01 | platinum (Pt) | 1.49E+07 | 2.19E+08 | tellurium (Te) | 4.60E+0 | 4 6.61E+05 |
| antimony concentrate | 2.15E+01 | 2.90E+02 | copper nitrate | 3.50E+02 | 4.98E+03 | iron sulfate | 4.14E+00 | 3.13E+01 | Pollack concentrate | 2.03E+01 | 2.73E+02 | terbium (Tb) | 3.47E+0 | 5 4.58E+06 |
| aqua regia | 8.50E+00 | 7.74E+01 | copper sulfate | 2.74E+01 | 3.07E+02 | isopropanol | 5.65E+00 | 6.18E+01 | polvaluminum chloride | 6.73E+00 | 5.72E+01 | tetrahvdrofuran | 1.19E+0 | 2 1.34E+03 |
| argon (Ar) | 2.20E+01 | 1.37E+04 | Copper-Chromium Catalysts | 4.28E+02 | 6.22E+03 | Itria. | 1.09E+04 | 1.43E+05 | Polyether compounds | 3.33E+03 | 4.90E+04 | thallium (TI) | 8.57E+0 | 2 1.21E+04 |
| arsenic (As) | 2.45E+02 | 3.29E+03 | cvanide disposal | 5.93E+02 | 4.77E+03 | kerosene | 3.38E+01 | 3.92E+02 | potassium chloride | 1.21E+02 | 1.57E+03 | thorium (Th) | 2.92E+0 | 2 4.38E+03 |
| arsenic concentrates | 5.58E+01 | 7.61E+02 | D2EHPA | 1.07E+04 | 1.57E+05 | krypton (Kr) | 6.15E+01 | 1.29E+04 | potassium hydroxide (KOH) | 2.13E+02 | 2.75E+03 | thorium concentrate | 1.63E+0 | 0 3.14E+01 |
| parite concentrate | 1.30E+01 | 1.37E+02 | defoamer | 9.68F+02 | 1 42F+04 | lauryl alcohol | 3 57E+01 | 6 23E+03 | praseodymium (Pr) | 3 47E+04 | 4 59E+05 | thulium oxide (TIO) | 8 90F+0 | 6 1 18E+08 |
| parium sulfate (BaSO4) (sulphat | 2 27E+01 | 2 42E+02 | deionized water | 5.04F-01 | 4 16E+00 | lead (the metal) | 1.86F+01 | 2 52E+02 | propylene | 2 08E+00 | 1.67E+01 | tin | 3 94E+0 | 3 5 81E+04 |
| pastne site | 1 16F+02 | 1.58E+03 | dichloromethylsilane | 9.88F+00 | 1.06F+02 | lead concentrate | 8.69E+00 | 1.32E+02 | propylene oxide | 1 92E+04 | 2.82E+05 | tinstone concentrate | 2 78F+0 | 2 4 10F+03 |
| pauxite | 3 49F-01 | 7 73E+00 | diethyl ether | 1 29F+01 | 1.40F+02 | limestone | 4 64F+00 | 7 06F+01 | quicklime | 8 72E+00 | 1.03E+02 | titanium (Ti) | 4 15E+0 | 2 4 64E+03 |
| enzene | 2.63E+01 | 4 27E+02 | dilute sulfuric acid | 2 68E+00 | 2 76E+01 | liquid nitrogen | 2.62E+00 | 8.61E+02 | Radioactive sludge treatment | 3.89E+01 | 3.97E+02 | toluene | 2.63E+0 | 0 1 70E+01 |
| pervl concentrate | 6.86F+01 | 1.04F+03 | dust treatment | 5 29F+00 | 5.44F+01 | lithium carbonate (Li2CO3) | 1 75E+01 | 1.87E+02 | Radioactive tailings processing | 1.24F+02 | 1.64E+03 | tributyl phosphorus | 9.82E+0 | 0 1 07E+02 |
| pervllium (Be) | 1 37E+03 | 1.83E+04 | dysprosium (Dy) | 8 54F+04 | 1 13E+06 | lithium chloride | 2 57E+02 | 3.07E+03 | radium (Ra) | 3.01E+05 | 4 29E+06 | tributylamine | 6.07E+0 | 1 6 83E+02 |
| pismuth (Bi) | 5 49E+01 | 7.51E+02 | erbium oxide (Erbium oxide) | 2 69E+05 | 3 55E+06 | Lithium concentrate | 3 48E+01 | 4 20F+02 | radon (Rn) | 6.73E+05 | 9.86F+06 | tributylphosphate | 5 16E+0 | 3 7 59E+04 |
| Bismuth concentrate | 1.44E+01 | 2.03E+02 | ethanol | 5.77E+00 | 3.21E+03 | lutetium oxide | 1.55E+07 | 2.05E+08 | Rare earth concentrates | 1.55E+03 | 2.05E+04 | Trioxide sulfate | 5.99E+0 | 0 6.79E+01 |
| oorax | 3 51E+01 | 4 89F+02 | ethylene oxide | 5 17E+00 | 3 75E+01 | Magnesite concentrate | 1 12F+01 | 1 40F+02 | red-light district | 1.38E+00 | 9 44F+01 | tungsten (W) | 1 18F+0 | 2 1 39E+03 |
| poric acid | 8 28F+01 | 1.08E+03 | ethylenediamine | 7 25E+00 | 7 77F+01 | magnesium (Mg) | 1.61F+02 | 1 75E+03 | rhenium (Re) | 5.57E+03 | 6.22E+04 | Tungsten concentrate | 1.80F+0 | 1 2 73E+02 |
| promine (Br) | 5 30E+03 | 1 20E+05 | ethylenediaminetetraacetic acid | 7 25E+00 | 7 77E+01 | magnesium chloride | 3.86F+01 | 4 41F+02 | rhodium (Rb) | 1 49E+07 | 2 19E+08 | uranium ore | 3.88E+0 | 2 5 69E+03 |
| outanol | 6 46E+03 | 9.49F+04 | europium oxide (EuO) | 1.08F+05 | 1.42E+06 | manganese (Mn) | 5.30E+01 | 5 40F+02 | rhodium catalyst | 2 71E+07 | 3 99E+08 | Uranium ore concentrate | 4 23E+0 | 2 6 04E+03 |
| putyl alcohol | 3 55E+01 | 3 99E+02 | Exhaust gas treatment | 7 14F-01 | 3 40F+00 | manganese concentrate | 1 27F+01 | 1.68E+02 | rubidium chloride | 8 24E+03 | 1 21E+05 | uranium oxide | 1 77E+0 | 1 2 34E+02 |
| outvraldehvde | 6 48E+03 | 9.51E+04 | Fe2O3 | 2.34F+01 | 3 43E+02 | Membrane Cleaner | 2 29E+01 | 2 16F+03 | Rubidium-bearing concentrates | 4 72E+02 | 6.93E+03 | V205 | 1.30E+0 | 2 1 89E+03 |
| cadmium (Cd) | 5 16E+04 | 8.32E+05 | fertilizer | 7 90F+00 | 7.60F+01 | mercury | 1 90F+02 | 2 64E+03 | ruthenium (Ru) | 4 70F+06 | 6.92E+07 | Vanadium concentrates | 7 29F+0 | 1 1 07E+03 |
| calcium carbide (CaC2) | 1.83E+01 | 2.08E+02 | flotation agent | 4 69E+00 | 9 30E+02 | methane | 9.23E+00 | 1 41F+02 | Salt Wastewater Treatment | 3.27E+00 | 3.26E+01 | | 5 24E+0 | 0 7 44F+01 |
| calcium chloride | 1.57E+01 | 1.84F+02 | fluorine (F) | 2 20E+03 | 2 45E+04 | methanol | 2 52E+00 | 4 91F+01 | salt water | 4 00F-01 | 1.81E+01 | waste acid treatment | 9 54F-0 | 1 5 88E+00 |
| calcium hydroxide (Ca(OH)2) | 7.31E+00 | 8 10F+01 | Fluorine-containing liquid treats | 4 66F+00 | 1.33E+02 | methyl isobutyl ketone | 6 75E+01 | 8 70F+02 | saltwater | 3.05E-01 | 6.89E+00 | Waste Alkali Treatment | 2 96F+0 | 0 3 10F+01 |
| alcium metal | 8.04E+01 | 8.81E+02 | foaming agent | 4.14E+01 | 4.46E+02 | molybdenum (Mo) | 5.25E+02 | 5.83E+03 | samarium (Sm) | 1.10E+05 | 1.46E+06 | Wastewater Treatment | 3.27E-0 | 1 1.33E+00 |
| calcium sulfate (CaSO4) (sulpha | 8 14F+00 | 9 92F+01 | formaldebyde | 6 58E+00 | 1.03E+02 | molybdenum concentrate | 2 41F+02 | 2 68E+03 | scandium (Sc) | 1.66F+04 | 2 25E+05 | water (esp. cool fresh water e | 0.00F+0 | 0 1 00F+00 |
| arbon dioxide | 4 15E+00 | 3 39E+01 | gadolinium oxide | 5 39E+04 | 7 11E+05 | money (written before an amou | 1.61E+07 | 2 41F+08 | selenium (Se) | 1.57E+04 | 2 26E+05 | vanthate | 4 15E+0 | 1 6 58E+02 |
| carbon disulfide | 1.08F+01 | 9.90F+01 | gallium (Ga) | 1.30E+03 | 1.39E+04 | natural gas | 8 58E+00 | 1.35E+02 | Silica concentrate | 1.07E+04 | 1.51E+02 | xenon (Xe) | 1 20E+0 | 2 1 87F+04 |
| carbon electrode | 2 06F+01 | 2 60F+02 | germanium (Ge) | 5.33E+03 | 8 50F+04 | Ne | 1.64F+02 | 2.31E+03 | silica sand | 9.74F+00 | 1 41F+02 | vtterbium oxide (YbO) | 1.55E+0 | 6 2 05E+07 |
| carbon monoxide | 1 14F+01 | 1.53E+02 | Glass matrix material | 3 91E+01 | 4 76E+02 | neodymium (Nd) | 7 94E+03 | 1.04F+05 | Silicone oil | 3 56E+01 | 3.85E+02 | zinc (Zn) | 1 97E+0 | 2 3 03E+03 |
| Celia | 3.88E+03 | 5 12E+04 | glucose | 1.88F+00 | 1.38E+03 | nickel (Ni) | 8 21F+02 | 1 15E+04 | silver | 2 71E+06 | 3 99E+07 | zinc concentrate | 7 28E+0 | 1 1 17E+03 |
| cement | 4.27E+00 | 2.72E+01 | gold concentrate | 6.07E+03 | 9.10E+04 | Nickel concentrate | 8.75E+01 | 1.25E+03 | silver concentrate | 3.83E+04 | 5.64E+05 | zircon sand | 4.18E+0 | 1 5.96E+02 |
| Ceres concentrates | 1.81E+01 | 2.65E+02 | hafnium (Hf) | 1.47E+03 | 1.76E+04 | niobium (Nb) | 3.01E+02 | 4.05E+03 | Sludge treatment | 9.29E-01 | 5.23E+00 | zirconium (Zr) | 1.72E+0 | 2 1.98E+03 |
| cesium chloride | 4.90E+03 | 7.11E+04 | HCN | 6.73E+00 | 7.42E+01 | niobium concentrate | 5.06E+01 | 6.37E+02 | sodium (Na) | 7.24E+01 | 7.58E+02 | | | |
| Cesium-containing concentrates | 1.98E+02 | 2.89E+03 | He | 2.41E+03 | 4.18E+04 | nitric acid | 3.61E+00 | 3.92E+01 | sodium carbonate (Na2CO3) | 2.28E+01 | 2.00E+02 | | | |
| | 1002.02 | 1.002.00 | | | 11102.104 | | | 1.022.01 | | _,_01 | 1002.02 | | | |

Total Material Requirement (TMR)

- **Definition**: TMR quantifies the total amount of materials extracted from nature to support an economy, including "hidden" material flows.
- **Purpose**: Developed to measure human reliance on natural resources and assess sustainability.

•Developed in the 1990s: Introduced as the "ecological rucksack" to represent the hidden environmental impacts of material extraction.

•Goal: Highlight the "invisible" resource usage and raise awareness of total environmental load.

•Pioneer of Footprint Concepts: Set the stage for various "footprint" metrics, focusing on resource dependency rather than direct environmental impact.



Evolution and Current Relevance of Total Material Requirement (TMR)

Legacy of the TMR Concept

•Originated at Germany's Wuppertal Institute and further developed by researchers like Halada from NIMS.

•TMR values were quantified and included as a supplementary indicator in Japan's Basic Environmental Plan.

•Shift in Focus from Environmental to Economic Factors

•Discussions on resource use and recycling moved towards availability and efficient use within the human economy.

•TMR became seen as one environmental factor among others, not directly linked to toxicological impacts, which led to reduced emphasis on it in policy and academic debates.

•Reduced Attention in the Era of Circular Economy

•Despite the rising importance of material intensity in circular economy discussions, the TMR concept has largely faded from focus, except among a few dedicated researchers.

TMR remains a valuable concept for understanding resource impact, though it has been overshadowed by broader circular economy perspectives as the resource edge point.

Positioning TMR in a Nature-Positive World



•Growing Importance of Nature-Positive Goals: As nature-positive approaches gain prominence, it's essential to assess and manage our true impact on natural resources, namely resource edge approach..

•**TMR as a Comprehensive Metric**: TMR captures the total material dependency, including hidden flows, providing a fuller picture of resource use beyond direct emissions or waste.

•Aligning TMR with Nature-Positive Actions: By reducing TMR, we directly support biodiversity, reduce habitat destruction, and promote sustainable resource management.

•Guiding Policy and Industry: TMR offers data for policies and business practices that align with nature-positive strategies, ensuring resources are used responsibly for a balanced, sustainable future.

Reasons Why TMR has been underutilized

•Ambiguity as an Environmental Indicator: TMR's role as an environmental load factor has been unclear, making it challenging to integrate into standard environmental impact assessments.

•Complex Data Requirements: Calculating TMR requires extensive data on hidden material flows, which was difficult to obtain and often incomplete.

 Limited Awareness and Focus: With more attention on metrics like carbon footprint, TMR was overshadowed and not widely adopted in sustainability practices.

•Lack of Standardization: Without standardized methods or frameworks, it was hard to apply TMR consistently across industries or regions.

•Data Accessibility: In the past, TMR calculations were reliant on limited data sources, but recent open data and AI advancements are changing this

Challenges in TMR Data Collection and the Role of Generative AI

- Challenges in Widespread Adoption of TMR
 - Data Complexity:.
 - Boundary Data Limitations:
- The Game Changer: Emergence of Generative AI (Large Language Models)
 - While often referred to as "generative," these AI systems fundamentally perform large-scale language processing grounded in extensive data mining.
 - Through careful prompting, these AI models can be guided to conduct targeted data mining, enabling access to previously inaccessible data, such as specific mining and waste management information on each specific site..

• New Potential with Data Mining AI

- By leveraging data mining capabilities in generative AI, we can obtain essential data for TMR calculations that was previously out of reach.
- This approach, termed Data Mining AI, represents a powerful tool for advancing TMR research and improving resource efficiency assessments.

Benefits of Using Generative AI for Data Mining in TMR

- Main Advantages of Generative AI in Data Mining
 - **Pattern and Trend Discovery**: Identifies hidden patterns and trends from large amounts of unstructured data, filling in details not covered by existing databases.
 - Real-Time Analysis: Processes global information in real-time, allowing rapid adaptation to new technologies and market shifts.
 - Cost-Effective Data Collection: Offers high-precision estimates even when field data collection is challenging, reducing costs and effort.

• Beyond Traditional Data Processing

- Generative AI surpasses traditional data processing limitations, providing integrated insights into complex systems and detailed processes including process-solvents and waste management.
- In TMR assessments for products with advanced technologies, Al-driven estimates offer detailed understanding of environmental impacts across production stages until resource edges.

Considerations and Challenges of Using Generative AI in LCA Data Mining

1. Accuracy of AI Predictions

•Generative AI relies on past data to make predictions, but accuracy heavily depends on data quality.

•Process data required for LCA varies greatly by industry, and tracking latest technologies and operations can be difficult.

•Human expertise is essential to validate AI predictions to ensure reliability.

2. Transparency and Interpretability

•AI models often act as "black boxes," making it hard to understand their decision-making process.

•LCA requires transparency and reproducibility, and lack of clarity in AI's outputs can undermine result credibility.

3. Potential for Bias

•AI predictions depend on the datasets it learns from; if biased, AI results may be skewed.

•This can lead to unfair outcomes for certain countries or companies, or underestimations of true environmental impacts.

Addressing Challenges in Using Generative AI for LCA Data Mining

•Balancing Opportunities and Challenges

Using generative AI for LCA data mining presents new opportunities but also introduces challenges related to accuracy, transparency, and bias.
It is crucial to understand AI's limitations and strengthen collaboration with human expertise to effectively address these challenges.

•Strategies for Effective AI Utilization

1.Setting Data Ranges for Uncertainty

Al provides not only a single estimate but also upper and lower bounds.
This allows LCA calculations to consider uncertainties, providing scenarios for both best and worst-case outcomes.

2.Expert Validation of AI Estimates

•Experts in LCA and production processes validate AI-derived data, discarding or adjusting any inappropriate results.

•Example: Specialists review AI's estimates for rare metal usage against historical data and current technology trends.

3.Reducing Bias Through Prompt Engineering

•The AI is fed diverse data sources to prevent reliance on a single viewpoint.

Enhancing Data Quality in LCA with Advanced Prompt Engineering for Generative AI

• Prompt Engineering Strategies for Generative AI

Specialist Perspective

- Specified AI as a "resource and materials engineering expert familiar with LCA, industry practices, and patents."
- Limited search scope to high-quality, specialized information, reducing the influence of general blogs or non-expert articles.

• Focusing on Established Industrial Methods

- Added prompts specifying "common industrial methods for obtaining target material."
- Ensured focus on commercially viable, practical data, reducing the risk of including lab-scale or speculative research data.

• Validation of Numerical Outputs

- When Al's estimates diverged significantly from expert knowledge, prompted AI to generate reaction names or equations.
- This helped confirm data relevance, reducing the chance of unrelated information mixing into AI outputs.

Outcome and Benefits

- Combining generative Al's capabilities with expert input and diversified data sources ensures high-quality, reliable data for CFP analysis.
- This approach enhances the accuracy of LCA results by managing uncertainties and mitigating bias, making it a robust tool for assessing complex materials.



Substance(s) input:.

- Natural Gas (Methane): Approximately 4.5 to 5.0 kg of methane is required.
- Water vapor (H₂O): Requires about 9 kg of water (used as steam).

Type and amount of energy:.

 It is mainly supplied as thermal energy; it takes 7.2-10 kWh of energy to produce 1 kg of hydrogen. Natural gas itself is also used as an energy source, emitting about 9-11 kg of CO₂ as a byproduct.

Solvents and their recirculation rates

 Water (steam) is used as a reactant, but the circulation rate is typically 0%. Water used in the reaction is consumed.

rust inhibitor

| Input Item Name | Suggested input amount |
|--|------------------------|
| Phosphoric acid (H ₃ PO ₄ , 85% concentration) | 0.4 to 0.5 kg |
| Zinc sulfate (ZnSO ₄ , 98% concentration) | 0.6 to 0.7 kg |
| Water (for reaction adjustment and dilution) | 0.1 to 0.2 kg |

| Solvents and other process substance s Name | Approximate amount used (kg) | Cycloparametric r ate |
|--|------------------------------|--------------------------|
| Catalyst (acidic or basic catalyst) | 0.005 to 0.01 kg | Almost 100%. |
| coolant | 5-10 kg | 80-90% (reused) |
| Solvent for cleaning (e.g., methanol) | 0.05 to 0.1 kg | 70-85% of the total |

| Type and amount of energy | Approximate consumption (kWh/kg) |
|---------------------------|----------------------------------|
| electric power | 0.2 to 0.4 kWh |

| element number Configurati (on Number Configuration Name factor weight element unit fuel cor energy Fuel CO2 fuel type Fuel quantity TMRofFu el CO2 emission s Builty 1 hydrogen 0 1 0 0 0 0 0 12 2 methane 4.743416 0 13.0444 natural ga 7.736901 0 2 methane 0 1 2.828427 0.194454 natural ga 0.70711 0.118844 0 2 methane 0 1 2.828427 0.194454 natural ga 0.070711 0.118844 0 4 natural gas 1.195826 0 <th>AD</th> | AD |
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| 4 | |
| 5 air 0.1 0 <td></td> | |
| 6 water (esp. cool, frei 1.549193 0 | |
| 7 amine solution 0.049497 0 | |
| 8 dimethyl ether 0.027111 0 | |
| 9 Sulfur Sludge Treatn 0.014142 0< | |
| 3 water vapor 0 1 0.022361 0.245967 natural ga 0.089443 0.144003 0 6 water (esp. cool, frest 1.098863 0 0 0 0 0 0 10 coolant 0.353553 0 0 0 0 0 0 0 11 rust inhibitor 0.01 0 <t< td=""><td></td></t<> | |
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| 10 coolant 0.353553 0 0 0 0 0 0 11 rust inhibitor 0.01 0 0 0 0 0 0 0 12 oxygen scavenger 0.01 0 0 0 0 0 0 0 4 natural gas oxygen scavenger 0.01 0 | |
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| 12 oxygen scavenger 0.01 0 | |
| 4 natural gas 0 0.999 0.387298 0.086963 natural ga 0.031623 0.050913 0 13 gas field gas 1.341641 0 | |
| 13 gas field gas 1.341641 0 | |
| 7 amine solution 0.00495 0 | |
| | |
| | |
| 15 methanol 0.003354 0 0 0 0 0 0 0 | |
| 6 water (esp. cool, fre: 0.632456 0 0 0 0 0 0 0 | |
| 5 air 1 0 1 0 0 0 0 0 0 | |
| 6 water (esp. cool, fres 0 1 0 0 0 0 0 0 | |
| 7 amine solution 0 1 0 0 0 0 0 0 0 | |
| 16 monoethanolamine 1 0 0 0 0 0 0 0 | |
| 8 dimethyl ether 0 1 0.282843 0.194454 natural ga 0.070711 0.113844 0 | |
| 15 | |

| А | Н | J | К | AQ | AR | AS | AT | AU | AV | AW | BR | BS | BT |
|-------------------|-----------------------------|------------------------|------------------------|-------------|----------|----------|---------------------|-------------|-----------------|----------|------------|----------|----------|
| element number | Configurat ion Number | element name | Configuration Name | Power FP | CO2FP | TCPF | Electricit y TMR | Fuel TMR | Resource TMR | TMR | | TCFP | TMR |
| 1 | | hydrogen | 0 | 41.69681 | 39.97782 | 59.95059 | 41.69681 | 12.39129 | 613.6438 | 704.8421 | hydrogen | 59 95059 | 704 8421 |
| | 2 | | methane | 39.93399 | 24.64637 | 43.77475 | 39.93399 | 10.31318 | 583.1306 | 668.919 | | | |
| | 3 | | water vapor | 1.762817 | 3.331453 | 4.175842 | 1.762817 | 2.078113 | 30.51319 | 35.92303 | | | |
| 2 | | methane | 0 | 8.418825 | 5.19591 | 9.228528 | 8.418825 | 2.174209 | 122.9347 | 141.0205 | methane | 9.228528 | 141.0205 |
| | 4 | | natural gas | 5.387193 | 4.580928 | 7.161393 | 5.387193 | 1.939441 | 115.6747 | 127.7959 | | | |
| | 5 | | air | 0 | 0 | 0 | 0 | 0 | 0.1 | 0.1 | | | |
| | 6 | | water (esp. cool, free | 0 | 0 | 0 | 0 | 0 | 1.549193 | 1.549193 | | | |
| | 7 | | amine solution | 0.140358 | 0.296227 | 0.363458 | 0.140358 | 0.103982 | 2.98971 | 3.358969 | | | |
| | 8 | | dimethyl ether | 0.052662 | 0.121061 | 0.146286 | 0.052662 | 0.015281 | 2.585581 | 2.700394 | | | |
| | 9 | | Sulfur Sludge Treatn | 0.010185 | 0.003241 | 0.008119 | 0.010185 | 0.00166 | 0.035552 | 0.056462 | | | |
| 3 | | water vapor | 0 | 0.195869 | 0.370161 | 0.463982 | 0.195869 | 0.230901 | 3.390355 | 3.991448 | water vap | 0.463982 | 3.991448 |
| | 6 | | water (esp. cool, free | 0 | 0 | 0 | 0 | 0 | 1.098863 | 1.098863 | | | |
| | 10 | | coolant | 0 | 0 | 0 | 0 | 0 | 0.353553 | 0.353553 | | | |
| | 11 | | rust inhibitor | 0.063231 | 0.050836 | 0.081124 | 0.063231 | 0.072898 | 1.095783 | 1.288188 | | | |
| | 12 | | oxygen scavenger | 0.110277 | 0.073358 | 0.126181 | 0.110277 | 0.014 | 0.842155 | 1.064579 | | | |
| 4 | | natural gas | 0 | 4.504997 | 3.830764 | 5.988658 | 4.504997 | 1.621842 | 96.73204 | 106.8683 | natural ga | 5.988658 | 106.8683 |
| | 13 | | gas field gas | 4.084938 | 3.689413 | 5.646098 | 4.084938 | 1.549102 | 95.46209 | 104.7317 | | | |
| | 7 | | amine solution | 0.014036 | 0.029623 | 0.036346 | 0.014036 | 0.010398 | 0.298971 | 0.335897 | | | |
| | 14 | | glycol | 0.012791 | 0.014856 | 0.020983 | 0.012791 | 0.009527 | 0.109969 | 0.143671 | | | |
| | 15 | | methanol | 0.002694 | 0.007156 | 0.008447 | 0.002694 | 0.000736 | 0.158997 | 0.164825 | | | |
| | 6 | | water (esp. cool, free | 0 | 0 | 0 | 0 | 0 | 0.632456 | 0.632456 | | | |
| 5 | | air | 0 | 0 | 0 | 0 | 0 | 0 | 1 | 1 | air | 0 | 1 |
| 6 | | water (esp. cool, free | 0 | 0 | 0 | 0 | 0 | 0 | 1 | 1 | water (esp | 0 | 1 |
| 7 | | amine solution | 0 | 2.835664 | 5.984684 | 7.342968 | 2.835664 | 2.100758 | 60.40125 | 67.86142 | amine sol | 7.342968 | 67.86142 |
| • | Sheet1 | (+) | | | | | • | | | | | | • |

| Excel | Sheet | for TMR | and | TCFP | | | | - | Japanese | • |
|--------|-------------------------|---------------|----------|-----------------|-------|--------------------------|-----------------------|---------------------------------------|-----------------------|-----|
| | | | | | | | | 1. A | | |
| The Ex | cel <mark>s</mark> heet | for calcula | tion of | TMR and TCFP | | | | | | |
| | | | | | | | | | | N |
| - 24 | Δ | E | н | 10 F | | к | | | | 100 |
| 1 e | lement nu | reference e C | onfigura | ti element name | Cont | figuration Name | factor weigł▲ | 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 | | |
| 2 | 1 | #DIV/0! | | | 0 | | 0 | 1. mar. | | |
| 3 | 2 | 3 | | 2 he | | | 0 | | | - |
| 4 | | 6 | | 3 | natu | iral gas | | | | |
| 5 | | 7 | | 4 | liqui | d nitrogen | | scat123plus | | |
| 6 | 3 | 4 | | natural gas | | | 0 | | | |
| 7 | 4 | 5 | | liquid nitrogen | | | 0 | | | |
| 8 | | 9 | | 5 | air | | | | | |
| 9 | 5 | 8 | | air | | | 0 | | | |
| 10 | 6 | 10 | | 6 Ne | | | 0 | | | |
| 11 | | 9 | | 5 | air | | | 100 A | | |
| 12 | | 13 | | 7 | nitro | gen (N) | | | | |
| 13 | 7 | 12 | | nitrogen (N) | | | 0 | 1.1.1 | | A |
| 14 | 8 | 14 | | 8 argon (Ar) | | | 0 | | | -P |
| 15 | | 9 | | 5 | air | | | | | 16 |
| 16 | | 7 | | 4 | liqui | d nitrogen | | | | |
| 17 | 9 | 17 | | 9 krypton (Kr) | | | 0 | | | |
| 18 | | 9 | | 5 | air | an 26 | | | | |
| 19 | 808 | 7 | | 4 | liqui | d nitrogen | 8 | SCAT123 | 0442 | |
| 20 | 10 | 20 | 1 | 0 xenon (Xe) | | | 0 | https://lca.ev | goods net/scat122/ | |
| 21 | | 9 | | 5 | air | an 26 | | 11103//100.50 | 150000.1161/ 0001120/ | |
| 22 | | 7 | 1 | 4 | liqui | d nitrogen | | | | |
| 23 | - | 24 | 1 | 1 | wate | er (esn. cool, fresh wat | er eø drinkinø√* ▶ | | | |
| 2 3 | = 5 | eet1 She | et2 + | | | | | | | |



☆ホーム / ■ TMR and TCFP

TMR and TCFP

TMR and TCFP

https://lca.sdgoods.net/tmr-tcfp/

All data for calculation

Common Excel sheet for calculation

| H TMR 7.05E2 TCFP 6.00E1 | | | Resource-view weight: tons of TMR for 1kg of metal production Total Carbon Footprint (from eco-sphere to a product) Total Material Requirements & Total CFP | | | | | | | | | | | | | | |
|--------------------------------------|-------------------------------|--------------------------|--|-------------------------------|--|--------------------------|------------------------|-------------------------|------------------------|------------------------|------------------------|--------------------------------|------------------------|------------------------|------------------------|------------------------|------------------------|
| Li L2CD3 1.87E2 1.75E1 | Be 1.83E4 1.37E3 | | Magnel Batteri | t motor | Ooti Info | cal function mation m | n eda 😐 | Display 4 Fire retar | its corish dant | Ins | | •B 8203 1.08E3 8.28E1 | C 3.52E2 2.74E1 | N 7.95E0 5.71E-1 | 0 9.99E0 6.38E-1 | F 2.45E4 2.20E3 | Ne 2.31E3 1.64E2 |
| Na NaOH 1.59E2 1.89E1 | Mg 1.75E3 1.61E2 | | IC tios and parts Thermoelectric. Solar cell Electric vining: Catabrist, electrode Isinthina: Structural material | | | | | | | | | | Si 8.12E2 7.25E1 | •р 1.44ЕЗ 1.06Е2 | S 6.86E1 6.06E0 | CI 1.76E2 2.08E1 | Ar 1.37E4 2.20E1 |
| KCI 1.57E3 1.21E2 | Ca CaO 1.03E2 8.72E0 | Sc 2.25E5 1.66E4 | • Ti 4.64E3 4.15E2 | V V2O5 1.89E3 1.30E2 | Cr 2 15E3 1.47E2 | Mn 5.40E2 5.30E1 | Fe 6.55E1 1.06E1 | Co 8.52E3 6.29E2 | Ni 1.15E4 8.21E2 | Cu 4.89E3 3.44E2 | Zn 3.03E3 1.97E2 | Ga 2.90E6 1.81E5 | Ge 8.50E4 5.33E3 | As 3.29E3 2.45E2 | Se 2.26E5 1.57E4 | Br 1.20E5 5.30E3 | Kr 1.29E4 6.15E1 |
| Rb RbCl 1.21E5 8.24E3 | SrCO3 7.29E2 5.79E1 | Y2O3 5.12E4 3.88E3 | Zr 1.98E3 1.72E2 | Nb 4.05E3 3.01E2 | Mo 2.68E3 2.41E2 | Tc | Ru 6.92E7 4.70E6 | Rh 2.19E8 1.49E7 | Pd 1.44E6 1.03E5 | Ag 3.99E7 2.71E8 | Cd 8.32E5 5.16E4 | e In 8.37E5 5.21E4 | Sn 5.81E4 3.94E3 | Sb 9.20E2 7.42E1 | Te 6.61E5 4.60E4 | 1 3.50E5 7.76E3 | Xe 1.87E4 1.20E2 |





scat123plus



COATION



Transition from CFP to TCFP for Comprehensive Resource Impact Analysis

• Why TCFP (Total Carbon Footprint) Instead of CFP

- TCFP extends calculations to the resource extraction phase, avoiding arbitrary boundary settings and cut-offs used in traditional CFP.
- This broader approach yields higher values, providing a more accurate assessment and reducing confusion with conventional, limited-scope CFP metrics.
- Traditional CFP mostly covers Scope 1 and 2 emissions of energy; TCFP captures Scope 3 of energy by tracking resources to their resource edge.
- This comprehensive approach overcomes limitations of arbitrary "system boundaries," providing a more accurate footprint

Footer: TCFP offers a full-spectrum view of environmental impact by tracing resource use from origin to final application, enhancing transparency and accuracy in carbon assessments..

Recalculated TMR

•Collaboration between data-mining AI and engineering experts provided TMR values for all elements and accepted chemicals, with open data and calculation basis.

Application to Total CFP

•Expanded system boundary to resource origins at the Earthhuman economy interface, overcoming "system boundary" limitations.

 Integrated process materials and waste treatment, often overlooked in traditional CFP.

•Extended energy calculations beyond Scope 1, accounting for impacts up to resource origin.

Outcome

•Transformed footprint calculations into an open data discussion, engaging wider audiences beyond LCA specialists. • Thank you! and Please open https://lca.sdgoods.net/tmr-tcfp/



I introduce you to a Open-data and Open Discussion System for TMR and TCFP !

Recycling issue: \rightarrow exchange efference flow

プロセス3から出たリサイクル物がブロセス4を経て、プロセス5とともに プロセス3に循環する場合

- B=∑pi*bi ← LCAの基本式
- =∑pi*bi+p3*b3+n4*Q'4*b4+(n2-p4/Q4)*Q5*b5
 i=1,2
- =S2+ p3*b3+p4*b4+(n2-m4)Q5*b5
- =∑pi*bi +(n2-m4)Q5*b5 1=1,4
- =S4+p3(1/Q'3-m'4)Q5*b5



静脈側の4は、供給ではなく処理プロセスとして3からの発生物処理となる。

供給として3は4を要求しない

3の単位操作の要求構成5の量として、単位操 作時の5への要求量から4処理プロセスから(2 の単位操作で)供給される分を引く。



Recursion problem



Then, change recrded H2 into "rH2" as another materal

And, Manually enter the value of H2 into the rH2 unit. If this changes the footprint value of H2, enter the new value.

Repeat this process until the error range is within the acceptable range.